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ORBITAL 9. In HF, it is assumed that the hydrogen 1s atomic orbital overlaps with a fluorine 2p orbital to form the bonding molecular orbital. The specific 2p orbital used in forming the bonding MO is the p orbital on the internuclear axis.

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johannaortega47. Chapter 9 Covalent Bonding. coordinate covalent bond. covalent bond. endothermic. exothermic. also known as a dative bond or coordinate bond is a kind of 2-.... also called a molecular bond, is a chemical

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bond that involves.... (of a reaction or process) accompanied by or requiring the abs....

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be able to distinguish between single, double, triple, sigma and pi bonds. 2. What determines the strength of covalent bonds? 3. Be able to draw Lewis structures (or electron dot diagrams) for Mg, Li, C, Al, S, Br<sub>2</sub>, MgO, HF, CH<sub>4</sub>, Li<sub>2</sub>S, and SO<sub>3</sub>. Name the structures and label ionic or covalent.

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molecule. a. chlorine b.

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